

Stereospecific Photochemical Rearrangements of Chiral Cyclohexenones

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Summary The stereospecificity of the photorearrangements of monocyclic chiral 4,4-disubstituted cyclohexenones has been studied; no racemization has been found in either the Type A or aryl migration pathways.

WE recently reported the photorearrangement of **(1a)** to **(2a)** and **(3a)** to be largely, if not totally, stereospecific.¹ The large limits of uncertainty (*ca.* 10%) were due to the small optical purity of **(1a)** and the method used to deter-

mine the optical purities [using the chiral ^1H n.m.r. shift reagent $\text{Eu}(\text{hfc})_3$], and represent the maximum reaction *via* non-synchronous mechanisms.² We now report the results of experiments using (1a) of higher optical purity, and (1b) of even greater optical purity. In the latter system, one observes aryl migration products, as well as Type A³ photoproducts, depending on the solvent.⁴ Some of the data obtained are shown in the Table.

TABLE

Compound	Optical rotation, $[\alpha]_D^{25}$	Optical purity, %
(1a) (starting)	(+) 11.93 \pm 0.01°	26.8 \pm 1.3
(1a) (recovered) ^a	(+) 11.89 \pm 0.03°	26.4 \pm 0.1
(2a) ^a	(-) 13.13 \pm 0.36°	27.2 \pm 1.0
(3a) ^a	(+) 9.96 \pm 0.50°	27.5 \pm 0.9
(4) ^a	(+) 7.81 \pm 0.07°	> 16.9 ^b
(1b) (starting)	(+) 53.83 \pm 0.02°	41.73 \pm 0.02
(1b) (recovered) ^c	(+) 51.25 \pm 2.38°	39.73 \pm 1.84 ^d
(2b) ^c	(-) 4.41 \pm 0.98°	41.9 \pm 1.5
(5) ^c	(-) 1.76 \pm 0.42°	41.2 \pm 1.3

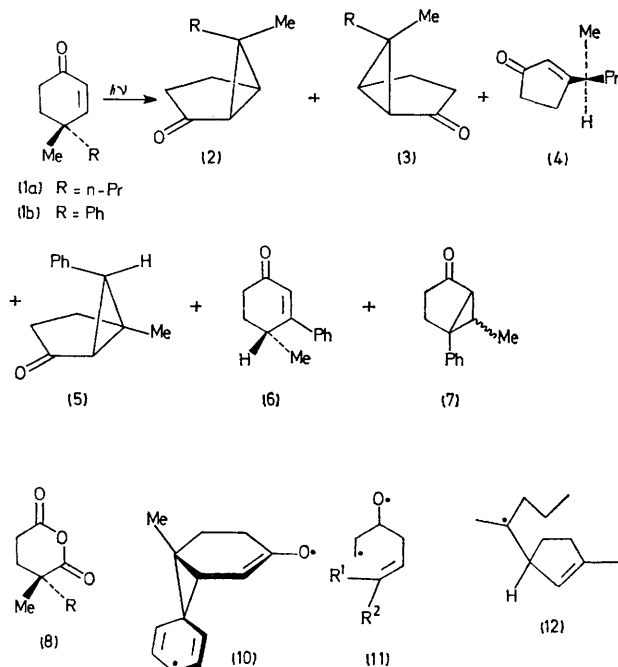
^a Photolysis in Bu^tOH for 12 h at 254 nm; ^b Minimum optical purity from oxidation to α -methylvaleric acid; ^c Photolysis in 95% EtOH for 24 h at >300 nm; ^d Large error due to small size of recovered sample. No racemization seen after 7 h photolysis.

The optical purity of (1b) was determined by polarimetry⁵ while that of (1a) was determined by oxidation to (8) and application of the ^1H n.m.r. chiral shift reagent technique.¹ Optical purities of (2), (3), and (5) were also determined using $\text{Eu}(\text{hfc})_3$. Gas-liquid partition chromatography was used to separate compounds, with multiple passes used when necessary to obtain pure material.

The absolute configurations of (1a) and (1b) were unambiguously assigned as *R*.⁵ The configurations of (2), (3a) and (5) were assigned using the 'inverse octant rule'⁶ obeyed by bicyclo[3.1.0]hexan-2-ones. The position of the substituents as *endo* or *exo* had been previously established.⁴ The configuration of (4) was established as *S* by Kuhn-Roth oxidation⁸ to (+)- α -methylvaleric acid (9) of known configuration.⁹ Experiments to determine the configuration of (6) are incomplete. Products (3b) and (7) from (1b)⁴ were not isolated in sufficient purity to permit stereochemical assignments.

Rearrangement of (1b) to (5) proceeds without loss of chirality and with no racemization of (1b). A possible intermediate in the formation of (5) and (6) is (10) which would lead to (6) with predicted configuration *R*. The rearrangements to (2) and (3) are at least 99% stereospecific.¹⁰ As postulated,¹ the rearrangement is specific on each face of the ring, giving diastereomeric lumiketones with inversion at the migrating carbon (C-4). Ring opened diradicals such as (11)¹¹ are no longer plausible as inter-

mediates in this reaction, which appears to be a $\sigma_2\text{a} + \pi_2\text{a}$ rearrangement.¹² This is despite the fact that the reaction originates from a triplet state⁷ (probably $^3\pi, \pi^*$)⁴ derived from excitation into either the $n \rightarrow \pi^*$ or $\pi \rightarrow \pi^*$ absorption bands of the carbonyl-containing chromophore.



Diradicals such as (12) have been proposed as intermediates in the formation of (4) from (1a).^{7,11a} Diradicals of this type account for epimerization of the lumiketones while maintaining configuration at C-3, and would thus lead to racemization on formation of lumiketones from (1). Since this is not the case,¹⁰ we conclude that such diradicals are not formed on photoexcitation of (1).^{1,10} Formation of (4) seems to involve in effect a $\sigma_2\text{a} + \sigma_2\text{a}$ cycloaddition mechanism, with simultaneous hydrogen shift and ring contraction.¹⁰

The quantum inefficiency in these rearrangements^{1,3,7} is probably associated with competitive crossing to the ground state potential surfaces of the starting material and the product from relaxed enone triplets with a highly twisted geometry.¹³

We thank Professor S. Weinreb (Fordham University) for helpful suggestions.

(Received, 5th September 1975; Com. 1010.)

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